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Introduction

The Idaho State Content Standards in Science are essential for developing the science literacy of Idaho students, as it is vital that our students understand the fundamental laws and practices within scientific disciplines. This document provides stakeholders with a set of rigorous and relevant science performance standards that prepare students to be informed, contributing citizens of the 21st-century world. The unifying goal is for Idaho students to practice and perform science and use their working knowledge of science to successfully function in a complex world.

The Idaho Science standards describe the knowledge and skills that students should learn, but they do not prescribe particular curricula, lessons, teaching techniques, or activities. Standards describe what students are expected to know and be able to do, while the local curriculum describes how teachers will help students master the standards. A wide variety of instructional resources may be used to meet the state content area standards. Decisions about curriculum and instruction are made locally by individual school districts and classroom teachers. The Idaho State Board of Education does not mandate the curriculum used within local schools. However, these science standards should be taught in a way that allows students to analyze the data and make their own decisions about what it means. Students should also be taught the current models and explanations of the scientific community regarding that data.

Organization and Structure of Science Standards

The Idaho Science Standards are based on *A Framework for K-12 Science Education*.¹ That framework outlines three organizational dimensions for each standard:

- **Science and Engineering Practices**
- **Disciplinary Core Ideas and Supporting Content**
- **Crosscutting Concepts**

The **Science and Engineering Practices** are used by students to demonstrate understanding of the disciplinary core ideas and crosscutting concepts. These practices update the scientific method in the classroom and include a wider range of skills for an expanded approach to how scientific investigations are conducted in the real world. Engaging in the practices of science and engineering helps students understand the wide range of approaches used to investigate natural phenomena and develop solutions to challenges. Students are expected to demonstrate grade appropriate proficiency in asking questions and defining problems; developing and using models; planning and carrying out investigations; analyzing and interpreting data; using mathematics and computational thinking; constructing explanations and designing solutions; engaging in argument from evidence; and obtaining, evaluating, and communicating information as they gather, analyze, and communicate scientific information.

The **Disciplinary Core Ideas and Supporting Content** are the focused, limited set of science ideas identified in the *Framework* as necessary for ALL students throughout their education and beyond their K–12 school years to achieve scientific literacy. The limited number of disciplinary core ideas

¹ National Research Council 2012. *A Framework for K-12 Science Education: Practices, Crosscutting Concepts, and Core Ideas*. Washington, DC: The National Academies Press, <https://doi.org/10.17226/13165>.

Instructional Shifts

While each standard incorporates the three dimensions, this alone does not drive student outcomes; ultimately, student learning depends on how the standards are translated to instructional practices. In order for students to attain the maximum benefit from the Idaho Science Standards, districts are encouraged to incorporate problem solving techniques and deep critical thinking exercises into those practices. Effective science teaching and learning integrates the three dimensions by allowing students to explain scientific phenomena, design solutions to real-world challenges, and build a foundation upon which they can continue to learn and to apply science knowledge and skills within and outside the K–12 education arena.

Interdisciplinary Approaches

The overlapping skills in the science and engineering practices, combined with the intellectual tools developed by the crosscutting concepts, build meaningful and substantive connections to interdisciplinary knowledge and skills in all content areas. Student understanding and retention are increased as connections are made to interdisciplinary learning which affords all students equitable access to learning and ensures all students are prepared for college, career, and citizenship.

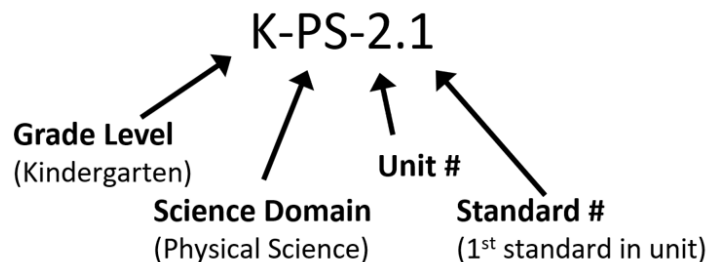
Using This Document

Each standard is followed by the Supporting Content (DCI) in order to add details of what knowledge should be mastered while students are working to achieve each standard. There are often *Further Explanations* and *Assessment Limits* following the content. The *Further Explanations* explain how the concepts embedded within the standard should be emphasized. These often contain examples that are not required, but give guidance about the complexity of ideas to ensure grade appropriate implementation. Assessment limits are similar and do not limit what content is learned about in the classroom, but they do keep assessments from expanding inappropriately outside of the grade-level expectations. These features are included when further clarity is needed.

The coding for each standard labels the grade level, science domain, unit number, and standard number as shown below:

Abbreviations

- K – Kindergarten
- MS – Middle School
- HS – High School
- LS – Life Science
- ESS – Earth and Space Science
- PS – Physical Science
- PSC – Physical Science Chemistry
- PSP – Physical Science Physics
- ETS – Engineering and Technology



K-ESS-2.2 Students who demonstrate understanding can:

Ask questions to obtain information about the purpose of weather forecasting to prepare for, and respond to, local weather.

Supporting Content ESS3.B: Natural Hazards

- Some kinds of weather are more likely than others in a given region. Weather scientists forecast the weather so that local communities can prepare for and respond to these events. (K-ESS-2.2)

Supporting Content ETS1.A: Delimiting an Engineering Problem

- Asking questions, making observations, and gathering information are helpful in thinking about problems. (K-ESS-2.2)

Further Explanation: Emphasis is on local forms of weather. Examples relating weather forecasting to preparing and responding could include using forecasts to plan for staying indoors during severe weather, going to cooling centers during heat waves, and covering windows before storms.

K-ESS-2.3 Students who demonstrate understanding can:

Communicate ideas that would enable humans to interact in a beneficial way with the land, water, air, and/or other living things in the local environment.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Things that people do can affect the world around them. People can reduce their effects on the land, water, air, and other living things. (K-ESS-2.3)

Supporting Content ETS1.B: Developing Possible Solutions

- Designs can be conveyed through sketches, drawings, or physical models. These representations are useful in communicating ideas for a problem's solutions to other people. (K-ESS-2.3)

Further Explanation: Examples of human influence on the land could include planting trees after a burn, protecting farm fields from erosion, or keeping plastic trash out of waterways.

1-LS-1.3 Students who demonstrate understanding can:

Use classification supported by evidence to differentiate between living and non-living items.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- Living and non-living things have distinct characteristics. (1-LS-1.3)

Further Explanation: Use a chart or Venn diagram to sort objects or pictures into living and non-living items.

1-LS-2 – Heredity: Inheritance and Variation of Traits

1-LS-2.1 Students who demonstrate understanding can:

Make observations to construct an evidence-based explanation that offspring are similar to, but not identical to, their parents.

Supporting Content LS3.A: Inheritance of Traits

- Young animals are very much, but not exactly, like their parents. Plants also are very much, but not exactly, like their parents. (1-LS-2.1)

Supporting Content LS3.B: Variation of Traits

- Individuals of the same kind of plant or animal are recognizable as similar but can also vary in many ways. (1-LS-2.1)

Further Explanation: Examples of patterns could include features plants or animals share. Examples of observations could include that leaves from the same kind of plant are the same shape but can differ in size, and that a particular breed of dog looks like its parents, but is not exactly the same.

Assessment Limit: Assessment does not include inheritance or animals that undergo metamorphosis or hybrids.

2-ESS-2 – Earth's Systems

2-ESS-2.1 Students who demonstrate understanding can:

Compare multiple solutions designed to slow or prevent wind or water from changing the shape of the land.

Supporting Content ESS2.A: Earth Materials and Systems

- Wind and water can change the shape of the land. (2-ESS-2.1)

Supporting Content ETS1.C: Optimizing the Design Solution

- Because there is always more than one possible solution to a problem, it is useful to compare and test designs. (2-ESS-2.1)

Further Explanation: Examples of solutions could include different designs of dikes and windbreaks to hold back wind and water, and different designs for using shrubs, grass, and trees to hold back the land.

2-ESS-2.2 Students who demonstrate understanding can:

Develop a model to represent the shapes and kinds of land and bodies of water in an area.

Supporting Content ESS2.B: Plate Tectonics and Large-Scale System Interactions

- Maps show where things are located. One can map the shapes and kinds of land and water in any area. (2-ESS-2.2)

Assessment Limit: Assessment does not include quantitative scaling in models.

2-ESS-2.3 Students who demonstrate understanding can:

Obtain information to identify where water is found on Earth and that it can be solid or liquid.

Supporting Content ESS2.C: The Roles of Water in Earth's Surface Processes

- Water is found in the ocean, rivers, lakes, and ponds. Water exists as solid ice and in liquid form. (2-ESS-2.3)

Grade 3

Physical Science

3-PS-1 – Motion and Stability: Forces and Interactions

3-PS-1.1 Students who demonstrate understanding can:

Plan and conduct an investigation to provide evidence of the effects of balanced and unbalanced forces on the motion of an object.

Supporting Content PS2.A: Forces and Motion

- Each force acts on one particular object with both strength and a direction. An object at rest typically has multiple forces acting on it, but they add to give zero net force on the object. Forces that do not sum to zero can cause changes in the object's speed or direction of motion. (Boundary: Qualitative and conceptual, but not quantitative, additions of forces are used at this level.) (3-PS-1.1)

Supporting Content PS2.B: Types of Interactions

- Objects in contact exert forces on each other (3-PS-1.1)

Further Explanation: Examples could include that an unbalanced force on one side of a ball can make it start moving; and that balanced forces pushing on a box from both sides will not produce any motion at all.

Assessment Limit: Assessment is limited to one variable at a time: number, size, or direction of forces. Assessment does not include quantitative force size, only qualitative and relative. Assessment is limited to gravity being addressed as a force that pulls objects down.

3-PS-1.2 Students who demonstrate understanding can:

Make observations and/or measurements of an object's motion to provide evidence that a pattern can be used to predict future motion.

Supporting Content PS2.A: Forces and Motion

- Force applied to an object can alter the position and motion of that object: revolve, rotate, float, sink, fall, and at rest. (3-PS-1.2)
- The patterns of an object's motion in various situations can be observed and measured; when that past motion exhibits a regular pattern, future motion can be predicted from it. (Boundary: Technical terms, such as magnitude, velocity, momentum, and vector quantity, are not introduced at this level, but the concept that some quantities need both size and direction to be described is developed.) (3-PS-1.2)

Further Explanation: Examples of motion with a predictable pattern could include a child swinging in a swing, a ball rolling back and forth in a bowl, and two children on a see-saw.

Assessment Limit: Assessment does not include technical terms such as period and frequency.

3-PS-1.3 Students who demonstrate understanding can:

Ask questions to determine cause and effect relationships of static electricity or magnetic interactions between two objects not in contact with each other.

Supporting Content PS2.B: Types of Interactions

- Electric and magnetic forces between a pair of objects do not require that the objects be in contact. The sizes of the forces in each situation depend on the properties of the objects and their distances apart. For forces between two magnets, the size of the force also depends on their orientation relative to each other. (3-PS-1.3, 3-PS-1.4)

Further Explanation: An example of static electricity force could include the force on hair from an electrically charged balloon. Examples of a magnetic force could include the force between two permanent magnets, the force between an electromagnet and steel paper clips, and the force exerted by one magnet versus the force exerted by two magnets. Examples of cause and effect relationships could include how the distance between objects affects strength of the force and how the orientation of magnets affects the direction of the magnetic force.

Assessment Limit: Assessment is limited to forces produced by objects that can be manipulated by students, and electrical interactions are limited to static electricity.

3-PS-1.4 Students who demonstrate understanding can:

Define a problem that can be solved by applying scientific ideas about magnets.

Supporting Content PS2.B: Types of Interactions

- Electric and magnetic forces between a pair of objects do not require that the objects be in contact. The sizes of the forces in each situation depend on the properties of the objects and their distances apart. For forces between two magnets, the size of the force also depends on their orientation relative to each other. (3-PS-1.3, 3-PS-1.4)

Further Explanation: Examples of problems could include constructing a latch to keep a door shut and creating a device to keep two moving objects from touching each other.

Life Science

3-LS-1 – From Molecules to Organisms: Structures and Processes

3-LS-1.1 Students who demonstrate understanding can:

Develop models to demonstrate that living things, although they have unique and diverse life cycles, all have birth, growth, reproduction, and death in common.

Supporting Content LS1.B: Growth and Development of Organisms

- Reproduction is essential to the continued existence of every kind of organism. Plants and animals have unique and diverse life cycles. (3-LS-1.1)

Further Explanation: Changes organisms go through during their life form a pattern.

Assessment Limit: Assessment of plant life cycles is limited to those of flowering plants. Assessment does not include details of human reproduction.

3-LS-2 – Ecosystems: Interactions, Energy, and Dynamics

3-LS-2.1 Students who demonstrate understanding can:

Construct an argument that some animals form groups that help members survive.

Supporting Content LS2.D: Social Interactions and Group Behavior

- Being part of a group helps animals obtain food, defend themselves, and cope with changes. Groups may serve different functions and vary dramatically in size. (3-LS-2.1)

3-LS-3 – Heredity: Inheritance and Variation of Traits

3-LS-3.1 Students who demonstrate understanding can:

Analyze and interpret data to provide evidence that plants and animals have traits inherited from parents and that variation of these traits exists in a group of similar organisms.

Supporting Content LS3.A: Inheritance of Traits

- Many characteristics of organisms are inherited from their parents. (3-LS-3.1)

Supporting Content LS3.B: Variation of Traits

- Different organisms vary in how they look and function because they have different inherited information. (3-LS-3.1)

Further Explanation: Patterns are the similarities and differences in traits shared between offspring and their parents, or among siblings. Emphasis is on non-human organisms.

Assessment Limit: Assessment does not include genetic mechanisms of inheritance and prediction of traits. Assessment is limited to non-human examples.

3-LS-3.2 Students who demonstrate understanding can:

Use evidence to support the explanation that traits can be influenced by the environment.

Supporting Content LS3.A: Inheritance of Traits

- Many characteristics involve both inheritance and environment. Characteristics result from individuals' interactions with the environment, which can range from diet to learning. (3-LS-3.2)

Supporting Content LS3.B: Variation of Traits

- The environment affects the traits that an organism develops. (3-LS-3.2)

Further Explanation: Examples of the environment affecting a trait could include that normally tall plants grown with insufficient water are stunted, and a pet dog that is given too much food and little exercise may become overweight.

3-LS-3.3 Students who demonstrate understanding can:

Construct an argument with evidence that in a particular habitat some organisms can survive well, some survive less well, and some cannot survive at all.

Supporting Content LS4.C: Adaptation

- For any particular environment, some kinds of organisms survive well, some survive less well, and some cannot survive at all. (3-LS-3.3)

Further Explanation: Examples of evidence could include needs, characteristics of the organisms, and habitats involved. The organisms and their habitat make up a system in which the parts depend on each other.

Earth and Space Science

3-ESS-1 – Earth’s Systems

3-ESS-1.1 Students who demonstrate understanding can:

Represent data in tables and graphical displays to describe typical weather conditions expected during a particular season.

Supporting Content ESS.D: Weather and Climate

- Scientists record patterns of the weather across different times and areas so that they can make predictions about what kind of weather might happen next. (3-ESS-1.1)

Further Explanation: Examples of data could include average temperature, precipitation, and wind direction.

Assessment Limit: Assessment of graphical displays is limited to pictographs and bar graphs. Assessment does not include climate change.

3-ESS-1.2 Students who demonstrate understanding can:

Obtain and combine information to describe climates in different regions of the world.

Supporting Content ESS.D: Weather and Climate

- Climate describes a range of an area's typical weather conditions and the extent to which those conditions vary over years. (3-ESS-1.2)

3-ESS-2 – Earth and Human Activity

3-ESS-2.1 Students who demonstrate understanding can:

Make a claim about the merit of a design solution that reduces the impacts of a weather-related hazard.

Supporting Content ESS3.B: Natural Hazards

- A variety of natural hazards result from natural processes. Humans cannot eliminate natural hazards but can take steps to reduce their impacts. (3-ESS-2.1)

Further Explanation: Examples of design solutions to weather-related hazards could include barriers to prevent flooding, wind-resistant roofs, and lightning rods.

Grade 4

Physical Science

4-PS-1 – Energy

4-PS-1.1 Students who demonstrate understanding can:

Use evidence to construct an explanation relating the speed of an object to the energy of that object.

Supporting Content PS3.A: Definitions of Energy

- The faster a given object is moving, the more energy it possesses. (4-PS-1.1)

Assessment Limit: Assessment does not include quantitative measures of changes in the speed of an object or any precise or quantitative definition of energy.

4-PS-1.2 Students who demonstrate understanding can:

Make observations to provide evidence that energy can be transferred by heat, sound, light, and electric currents.

Supporting Content PS3.A: Definitions of Energy

- Energy can be moved from place to place by moving objects or through heat, sound, light, or electric currents. (4-PS-1.2, 4-PS-1.3)

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- Energy is present whenever there are moving objects, sound, light, or heat. When objects collide, energy can be transferred from one object to another, thereby changing their motion. In such collisions, some energy is typically also transferred to the surrounding air; as a result, the air gets heated and sound is produced. (4-PS-1.2, 4-PS-1.3)
- Light transfers energy from place to place. (4-PS-1.2)
- Energy can be transferred from place to place by electric currents, which can then be used locally to produce motion, sound, heat, or light. The currents may have been produced by transforming the energy of motion into electrical energy. (4-PS-1.2, 4-PS-1.4)

Assessment Limit: Assessment does not include quantitative measurements of energy.

4-PS-1.3 Students who demonstrate understanding can:

Ask questions and predict outcomes about the changes in energy that occur when objects collide.

Supporting Content PS3.A: Definitions of Energy

- Energy can be moved from place to place by moving objects or through heat, sound, light, or electric currents. (4-PS-1.2, 4-PS-1.3)

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- Energy is present whenever there are moving objects, sound, light, or heat. When objects collide, energy can be transferred from one object to another, thereby changing their motion. In such collisions, some energy is typically also transferred to the surrounding air; as a result, the air gets heated and sound is produced. (4-PS-1.2, 4-PS-1.3)

Supporting Content PS3.C: Relationship Between Energy and Forces

- When objects collide, the contact forces transfer energy so as to change the objects' motions. (4-PS-1.3)

Further Explanation: Emphasis is on the change in the energy due to the change in speed, not on the forces, as objects interact.

Assessment Limit: Assessment does not include quantitative measurements of energy.

4-PS-1.4 Students who demonstrate understanding can:

Apply scientific ideas to design, test, and refine a device that converts energy from one form to another.

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- Energy can be transferred from place to place by electric currents, which can then be used locally to produce motion, sound, heat, or light. The currents may have been produced by transforming the energy of motion into electrical energy. (4-PS-1.2, 4-PS-1.4)

Supporting Content PS3.D: Energy in Chemical Processes and Everyday Life

- The expression “produce energy” typically refers to the conversion of stored energy into a desired form for practical use. (4-PS-1.4)

Supporting Content ETS1.A: Defining Engineering Problems

- Possible solutions to a problem are limited by available materials and resources (constraints). The success of a designed solution is determined by considering the desired features of a solution (criteria). Different proposals for solutions can be compared on the basis of how well each one meets the specified criteria for success or how well each takes the constraints into account. (4-PS-1.4)

Further Explanation: Examples of devices could include electric circuits that convert electrical energy into motion energy of a vehicle, light, or sound, and a passive solar heater that converts light into heat. Examples of constraints could include the materials, cost, or time to design the device.

Assessment Limit: Devices should be limited to those that convert motion energy to electric energy or use stored energy to cause motion or produce light or sound.

4-PS-2 – Waves

4-PS-2.1 Students who demonstrate understanding can:

Develop a model of a simple mechanical wave to describe patterns of amplitude and wavelength and that waves can cause objects to move.

Supporting Content PS4.A: Wave Properties

- Waves, which are regular patterns of motion, can be made in water by disturbing the surface. When waves move across the surface of deep water, the water goes up and down in place; there is no net motion in the direction of the wave except when the water meets a beach. (4-PS-2.1)
- Waves of the same type can differ in amplitude (height of the wave) and wavelength (spacing between wave peaks). (4-PS-2.1)

Further Explanation: Examples of models could include diagrams, analogies, and physical models using wire to illustrate wavelength and amplitude of waves.

Assessment Limit: Assessment does not include interference effects, electromagnetic waves, non-periodic waves, or quantitative models of amplitude and wavelength.

4-PS-2.2 Students who demonstrate understanding can:

Develop a model to describe that light reflecting from objects and entering the eye allows objects to be seen.

Supporting Content PS4.B: Electromagnetic Radiation

- An object can be seen when light reflected from its surface enters the eyes. (4-PS-2.2)

Assessment Limit: Assessment does not include knowledge of specific colors reflected and seen, the cellular mechanisms of vision, or how the retina works.

4-PS-2.3 Students who demonstrate understanding can:

Generate and compare multiple solutions that use patterns to transfer information.

Supporting Content PS4.C: Information Technologies and Instrumentation

- Digitized information can be transmitted over long distances without significant degradation. High-tech devices, such as computers or cell phones, can receive and decode information—convert it from digitized form to voice—and vice versa. (4-PS-2.3)

Supporting Content ETS1.C: Optimizing the Design Solution

- Different solutions need to be tested in order to determine which of them best solves the problem, given the criteria and the constraints. (4-PS-2.3)

Further Explanation: Examples of solutions could include drums sending coded information through sound waves, using a grid of 1s and 0s representing black and white to send information about a picture, and using Morse code to send text.

Life Science

4-LS-1 – From Molecules to Organisms: Structures and Processes

4-LS-1.1 Students who demonstrate understanding can:

Construct an argument that plants and animals have internal and external structures that function to support survival, growth, behavior, and reproduction.

Supporting Content LS1.A: Structure and Function

- Plants and animals have both internal and external structures that serve various functions in growth, survival, behavior, and reproduction. (4-LS-1.1)
- Animals have various body systems with specific functions for sustaining life: skeletal, circulatory, respiratory, muscular, digestive, etc. (4-LS-1.1)

Further Explanation: Examples of structures could include thorns, stems, roots, colored petals, heart, stomach, lung, brain, and skin.

Assessment Limit: Assessment is limited to macroscopic structures within plant and animal systems.

4-LS-1.2 Students who demonstrate understanding can:

Use a model to describe how animals receive different types of information through their senses, process the information in their brain, and respond to the information in different ways.

Supporting Content LS1.D: Information Processing

- Different sense receptors are specialized for particular kinds of information, which may be then processed by the animal's brain. Animals are able to use their perceptions and memories to guide their actions. (4-LS-1.2)

Further Explanation: Emphasis is on systems of information transfer.

Assessment Limit: Assessment does not include the mechanisms by which the brain stores and recalls information or the mechanisms of how sensory receptors function.

Earth and Space Science

4-ESS-1 – Earth’s Place in the Universe

4-ESS-1.1 Students who demonstrate understanding can:

Identify evidence from patterns in rock formations and fossils in rock layers for changes in a landscape over time to support an explanation for changes in a landscape over time.

Supporting Content ESS1.C: The History of Planet Earth

- Local, regional, and global patterns of rock formations reveal changes over time due to earth forces, such as earthquakes. The presence and location of certain fossil types indicate the order in which rock layers were formed. (4-ESS-1.1)
- There are three classifications of rocks produced within the rock cycle: sedimentary, metamorphic, and igneous. (4-ESS-1.1)

Further Explanation: Examples of evidence from patterns could include rock layers with marine shell fossils above rock layers with plant fossils and no shells, indicating a change from land to water over time; and a canyon with different rock layers in the walls and a river in the bottom, indicating that over time a river cut through the rock.

Assessment Limit: Assessment does not include specific knowledge of the mechanism of rock formation or memorization of specific rock formations and layers. Assessment is limited to relative time.

4-ESS-2 – Earth’s Systems

4-ESS-2.1 Students who demonstrate understanding can:

Make observations and/or measurements to provide evidence of the effects of weathering or the rate of erosion by water, ice, wind, or vegetation.

Supporting Content ESS2.A: Earth Materials and Systems

- Rainfall helps to shape the land and affects the types of living things found in a region. Water, ice, wind, living organisms, and gravity break rocks, soils, and sediments into smaller particles and move them around. (4-ESS-2.1)

Supporting Content ESS2.E: Biogeology

- Living things affect the physical characteristics of their regions. Examples could include a beaver constructing a dam to create a pond or tree roots breaking a rock. (4-ESS-2.1)

Further Explanation: Examples of variables to test could include angle of slope in the downhill movement of water, amount of vegetation, speed of wind, relative rate of deposition, cycles of freezing and thawing of water, cycles of heating and cooling, and volume of water flow.

Assessment Limit: Assessment is limited to a single form of weathering or erosion.

4-ESS-2.2 Students who demonstrate understanding can:

Analyze and interpret data from maps to describe patterns of Earth’s features.

Supporting Content ESS2.B: Plate Tectonics and Large-Scale System Interactions

- The locations of mountain ranges, deep ocean trenches, ocean floor structures, earthquakes, and volcanoes occur in patterns. Most earthquakes and volcanoes occur in bands that are often along the boundaries between continents and oceans. Major mountain chains form inside continents or near their edges. Maps can help locate the different land and water features in different areas of Earth. (4-ESS-2.2)

Further Explanation: Maps can include topographic maps of Earth’s land and ocean floor, as well as maps of the locations of mountains, continental boundaries, volcanoes, and earthquakes.

4-ESS-3 – Earth and Human Activity

4-ESS-3.1 Students who demonstrate understanding can:

Obtain and combine information to describe that energy and fuels are derived from natural resources and their uses affect the environment.

Supporting Content ESS3.A: Natural Resources

- Energy and fuels that are modified from natural sources affect the environment in multiple ways. Some resources are renewable over time, and others are not. (4-ESS-3.1)

Further Explanation: Examples of renewable energy resources could include wind energy, water behind dams, and sunlight; non-renewable energy resources are fossil fuels and atomic energy. Examples of environmental effects could include biological effects from moving parts, erosion, change of habitat, and pollution.

4-ESS-3.2 Students who demonstrate understanding can:

Generate and compare multiple solutions to reduce the impacts of natural Earth processes on humans.

Supporting Content ESS3.B: Natural Hazards

- A variety of hazards result from natural processes (e.g., earthquakes, floods, tsunamis, volcanic eruptions). Hazards cannot be eliminated, but their impacts can be reduced. (4-ESS-3.2)

Supporting Content ETS1.B: Designing Solutions to Engineering Problems

- Testing a solution involves investigating how well it performs under a range of likely conditions. (4-ESS-3.2)

Further Explanation: Examples of solutions could include designing an earthquake-resistant building and improving monitoring of volcanic activity.

Assessment Limit: Assessment is limited to earthquakes, floods, tsunamis, and volcanic eruptions.

5-PS-1.4 Students who demonstrate understanding can:

Conduct an investigation to determine whether the mixing of two or more substances results in new substances.

Supporting Content PS1.B: Chemical Reactions

- When two or more different substances are mixed, a new substance with different properties may be formed. (5-PS-1.4)

5-PS-2 – Motion and Stability: Forces and Interactions

5-PS-2.1 Students who demonstrate understanding can:

Support an argument that Earth’s gravitational force exerted on objects is directed downward.

Supporting Content PS2.B: Types of Interactions

- The gravitational force of Earth acting on an object near Earth’s surface pulls that object toward the planet’s center. (5-PS-2.1)

Further Explanation: “Downward” is a local description of the direction that points toward the center of the spherical Earth. Examples could include reasoning that since an object that is initially stationary when held moves downward when it is released, there must be a force (gravity) acting on the object that pulls the object toward the center of Earth.

Assessment Limit: Assessment does not include mathematical representation of gravitational force.

5-PS-3 – Energy

5-PS-3.1 Students who demonstrate understanding can:

Use models to describe that energy in animals’ food (used for body repair, growth, motion, and to maintain body warmth) was once energy from the Sun.

Supporting Content PS3.D: Energy in Chemical Processes and Everyday Life

- The energy released from food was once energy from the Sun. The energy was captured by plants in the chemical process that forms plant matter (from air and water). (5-PS-3.1, 5-LS-1.1)

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- Food provides animals with the materials they need for body repair and growth and the energy they need to maintain body warmth and for motion. (5-PS-3.1)

Further Explanation: Examples of models could include diagrams and flow charts.

5-LS-2.3 Students who demonstrate understanding can:

Make a claim about the merit of a solution to a problem caused when the environment changes and the types of plants and animals living there may change.

Supporting Content LS2.C: Ecosystem Dynamics, Functioning, and Resilience

- When the environment changes in ways that affect a place's physical characteristics, temperature, or availability of resources, some organisms survive and reproduce, others move to new locations, yet others move into the transformed environment, and some die. (5-LS-2.3)

Supporting Content LS4.D: Biodiversity

- Populations live in a variety of habitats, and change in those habitats affects the organisms living there. (5-LS-2.3)

Further Explanation: Examples of environmental changes could include changes in land characteristics, water distribution, temperature, food, and other organisms.

Assessment Limit: Assessment is limited to a single environmental change. Assessment does not include the greenhouse effect or climate change.

5-LS-2.4 Students who demonstrate understanding can:

Develop a model to describe the movement of matter among plants, animals, decomposers, and the environment.

Supporting Content LS2.A: Interdependent Relationships in Ecosystems

- The food of almost any kind of animal can be traced back to plants. Organisms are related in food webs in which some animals eat plants for food and other animals eat the animals that eat plants. Some organisms, such as fungi and bacteria, break down dead organisms (both plants and animals) and therefore operate as "decomposers." Decomposition eventually restores (recycles) some materials back to the soil. Organisms can survive only in environments in which their particular needs are met. A healthy ecosystem is one in which multiple species of different types are each able to meet their needs in a relatively stable web of life. Newly introduced species can damage the balance of an ecosystem. (5-LS-2.4)

Supporting Content LS2.B: Cycles of Matter and Energy Transfer in Ecosystems

- Matter cycles between the air and soil, and among plants, animals, and microbes as these organisms live and die. Organisms obtain gases and water, from the environment, and release waste matter (gas, liquid, or solid) back into the environment. (5-LS-2.4)

Further Explanation: Emphasis is on the idea that matter that is not food (air, water, decomposed materials in soil) is changed by plants into matter that is food. Examples of systems could include organisms, ecosystems, and the Earth.

Assessment Limit: Assessment does not include molecular explanations.

5-ESS-2.2 Students who demonstrate understanding can:

Describe and graph the relative amounts of fresh and salt water in various reservoirs, to interpret and analyze the distribution of water on Earth.

Supporting Content ESS2.C: The Roles of Water in Earth's Processes

- Nearly all of Earth's available water is in the ocean. Most freshwater is in glaciers or underground; only a tiny fraction is in streams, lakes, wetlands, and the atmosphere. (5-ESS-2.2)

Assessment Limit: Assessment is limited to oceans, lakes, rivers, glaciers, groundwater, and polar ice caps, and does not include the atmosphere.

5-ESS-3 – Earth and Human Activity

5-ESS-3.1 Students who demonstrate understanding can:

Obtain and combine information about ways communities protect Earth's resources and environment using scientific ideas.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Human activities in agriculture, industry, and everyday life have effects on the land, vegetation, streams, ocean, air, and even outer space. Individuals and communities can often mitigate these effects through innovation and technology. (5-ESS-3.1)

MS-PS-1.3 Students who demonstrate understanding can:

Construct a scientific explanation, based on evidence, to describe that synthetic materials come from natural resources.

Supporting Content PS1.A: Structure and Properties of Matter

- Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it. (MS-PS-1.2, MS-PS-1.3)

Supporting Content PS1.B: Chemical Reactions

- Substances react chemically in characteristic ways. In a chemical process, the atoms that make up the original substances are regrouped into different molecules, and these new substances have different properties from those of the reactants. (MS-PS-1.2, MS-PS-1.3, MS-PS-1.5)

Further Explanation: Emphasis is on natural resources that undergo a chemical process to form the synthetic material. Examples of new materials could include new medicine, foods, plastics, and alternative fuels.

Assessment Limit: Assessment is limited to qualitative information.

MS-PS-1.4 Students who demonstrate understanding can:

Develop a model that predicts and describes changes in particle motion, temperature, and state of a pure substance when thermal energy is added or removed.

Supporting Content PS1.A: Structure and Properties of Matter

- Gases and liquids are made of molecules or inert atoms that are moving about relative to each other. (MS-PS-1.4)
- In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely spaced and may vibrate in position but do not change relative locations. (MS-PS-1.4)
- The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter. (MS-PS-1.4)

Supporting Content PS3.A: Definitions of Energy

- The term “heat” as used in everyday language refers both to thermal energy (the motion of atoms or molecules within a substance) and the transfer of that thermal energy from one object to another. In science, heat is used only for this second meaning; it refers to the energy transferred due to the temperature difference between two objects. (MS-PS-1.4)
- The temperature of a system is proportional to the average internal kinetic energy and potential energy per atom or molecule (whichever is the appropriate building block for the system’s material). The details of that relationship depend on the type of atom or molecule and the interactions among the atoms in the material. Temperature is not a direct measure of a system’s total thermal energy. The total thermal energy (sometimes called total internal energy) of a system depends jointly on the temperature, the total number of atoms in the system, and the state of the material. (MS-PS-1.4)

Further Explanation: Emphasis is on qualitative molecular-level models of solids, liquids, and gases to show that adding or removing thermal energy increases or decreases kinetic energy of the particles until a change of state occurs. Examples of models could include drawings and diagrams. Examples of particles could include molecules or inert atoms. Examples of pure substances could include water, carbon dioxide, and helium.

MS-PS-1.5 Students who demonstrate understanding can:

Develop and use a model to describe how the total number of atoms does not change in a chemical reaction and thus mass is conserved.

Supporting Content PS1.B: Chemical Reactions

- Substances react chemically in characteristic ways. In a chemical process, the atoms that make up the original substances are regrouped into different molecules, and these new substances have different properties from those of the reactants. (MS-PS-1.2, MS-PS-1.3, MS-PS-1.5)
- The total number of each type of atom is conserved, and thus the mass does not change. (MS-PS-1.5)

Further Explanation: Emphasis is on law of conservation of matter and on physical models or drawings, including digital forms, that represent atoms.

Assessment Limit: Assessment does not include the use of atomic masses, balancing symbolic equations, or intermolecular forces.

MS-PS-1.6 Students who demonstrate understanding can:

Undertake a design project to construct, test, and/or modify a device that either releases or absorbs thermal energy by chemical processes.

Supporting Content PS1.B: Chemical Reactions

- Some chemical reactions release energy, others store energy. (MS-PS-1.6)

Supporting Content ETS1.B: Developing Possible Solutions

- A solution needs to be tested and then modified on the basis of the test results in order to improve it. (MS-PS-1.6)

Supporting Content ETS1.C: Developing Possible Solutions

- Although one design may not perform the best across all tests, identifying the characteristics of the design that performed the best in each test can provide useful information for the redesign process — that is, some of the characteristics may be incorporated into the new design. (MS-PS-1.6)
- The iterative process of testing the most promising solutions and modifying what is proposed on the basis of the test results leads to greater refinement and ultimately to an optimal solution. (MS-PS-1.6)

Further Explanation: Emphasis is on the design, controlling the transfer of energy to the environment, and modification of a device using factors such as type and concentration of a substance. Examples of designs could involve chemical reactions such as dissolving ammonium chloride or calcium chloride (i.e., hand-warmers).

Assessment Limit: Assessment is limited to the criteria of amount, time, and temperature of substance in testing the device.

MS-PS-2.4 Students who demonstrate understanding can:

Construct and present arguments using evidence to support the claim that gravitational interactions are attractive and depend on the masses of interacting objects.

Supporting Content PS2.B: Types of Interactions

- Gravitational forces are always attractive. There is a gravitational force between any two masses, but it is very small except when one or both of the objects have large mass—e.g., Earth and the Sun. (MS-PS-2.4)

Further Explanation: Examples of evidence for arguments could include data generated from simulations or digital tools, and charts displaying mass, strength of interaction, distance from the Sun, and orbital periods of objects within the solar system.

Assessment Limit: Assessment does not include Newton’s Law of Gravitation or Kepler’s Laws.

MS-PS-2.5 Students who demonstrate understanding can:

Conduct an investigation and evaluate the experimental design to provide evidence that fields exist between objects exerting forces on each other even though the objects are not in contact.

Supporting Content PS2.B: Types of Interactions

- Forces that act at a distance (electric, magnetic, and gravitational) can be explained by fields that extend through space and can be mapped by their effect on a test object (a charged object, or a ball, respectively). (MS-PS-2.5)

Further Explanation: Examples of this phenomenon could include the interactions of magnets, electrically charged strips of tape, and electrically charged pith balls. Examples of investigations could include first-hand experiences or simulations.

Assessment Limit: Assessment is limited to electric and magnetic fields, and limited to qualitative evidence for the existence of fields.

- A solution needs to be tested, and then modified on the basis of the test results in order to improve it. There are systematic processes for evaluating solutions with respect to how well they meet criteria and constraints of a problem. (MS-PS-3.3)

Further Explanation: Examples of devices could include an insulated box, a solar cooker, and a Styrofoam cup.

Assessment Limit: Assessment does not include calculating the total amount of thermal energy transferred.

MS-PS-3.4 Students who demonstrate understanding can:

Plan an investigation to determine the relationships among the energy transferred, the type of matter, the mass, and the change in the average kinetic energy of the particles as measured by the temperature of the sample.

Supporting Content PS3.A: Definitions of Energy

- Temperature is a measure of the average kinetic energy of particles of matter. The relationship between the temperature and the total energy of a system depends on the types, states, and amounts of matter present. (MS-PS-3.3, MS-PS-3.4)

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- The amount of energy transfer needed to change the temperature of a matter sample by a given amount depends on the nature of the matter, the size of the sample, and the environment. (MS-PS-3.4)

Further Explanation: Examples of experiments could include comparing final water temperatures after different masses of ice melted in the same volume of water with the same initial temperature, the temperature change of samples of different materials with the same mass as they cool or heat in the environment, or the final temperature of samples of the same material with different masses when a specific amount of energy is added.

Assessment Limit: Assessment does not include calculating the total amount of thermal energy transferred.

MS-PS-3.5 Students who demonstrate understanding can:

Construct, use, and present arguments to support the claim that when the kinetic energy of an object changes, energy is transferred to or from the object.

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- When the motion energy of an object changes, there is inevitably some other change in energy at the same time. (MS-PS-3.5)

Further Explanation: Examples of empirical evidence used in arguments could include an inventory or other representation of the energy before and after the transfer in the form of temperature changes or motion of object.

Assessment Limit: Assessment does not include calculations of energy.

Life Science

MS-LS-1 – From Molecules to Organisms: Structure and Processes

MS-LS-1.1 Students who demonstrate understanding can:

Conduct an investigation to provide evidence that living things are made of cells; either one cell or many different numbers and types of cells.

Supporting Content LS1.A: Structure and Function

- All living things are made up of cells, which is the smallest unit that can be said to be alive. An organism may consist of one single cell (unicellular) or many different numbers and types of cells (multicellular). (MS-LS-1.1)

Further Explanation: Emphasis is on developing evidence that living things are made of cells, distinguishing between living and non-living things, and understanding that living things may be made of one cell or many and varied cells.

MS-LS-1.2 Students who demonstrate understanding can:

Develop and use a model to describe the function of a cell as a whole and ways parts of cells contribute to the function.

Supporting Content LS1.A: Structure and Function

- Within cells, special structures are responsible for particular functions, and the cell membrane forms the boundary that controls what enters and leaves the cell. (MS-LS-1.2)

Further Explanation: Emphasis is on the cell functioning as a whole system and the primary role of identified parts of the cell, specifically the nucleus, chloroplasts, mitochondria, cell membrane, and cell wall. These are visible with a light microscope.

Assessment Limit: Assessment of organelle structure/function relationships is limited to the cell wall and cell membrane. Assessment of the function of the other organelles is limited to their relationship to the whole cell. Assessment does not include the biochemical function of cells or cell parts.

MS-LS-1.3 Students who demonstrate understanding can:

Make a claim supported by evidence for how a living organism is a system of interacting subsystems composed of groups of cells.

Supporting Content LS1.A: Structure and Function

- In multicellular organisms, the body is a system of multiple interacting subsystems. These subsystems are groups of cells that work together to form tissues and organs that are specialized for particular body functions. (MS-LS-1.3)

Further Explanation: Emphasis is on the conceptual understanding that cells form tissues and tissues form organs specialized for particular body functions. Examples could include the interaction of subsystems within a system and the normal functioning of those systems.

Assessment Limit: Assessment does not include the mechanism of one body system independent of others. Assessment is not focused on human body systems.

MS-LS-1.4 Students who demonstrate understanding can:

Construct a scientific argument based on evidence to defend a claim of life for a specific object or organism.

Supporting Content LS1.B: Characteristics of Living Things

- Organisms reproduce, either sexually or asexually, and transfer their genetic information to their offspring. (MS-LS-1.4)
- Living things share certain characteristics. (These include response to environment, reproduction, energy use, growth and development, life cycles, made of cells, etc.) (MS-LS-1.4)

Further Explanation: Examples should include both biotic and abiotic items, and should be defended using accepted characteristics of life.

Assessment Limit: Assessment does not include specific conclusions regarding the living status of viruses, or other disputed examples.

MS-LS-1.5 Students who demonstrate understanding can:

Construct a scientific explanation based on evidence for the role of photosynthesis in the cycling of matter and flow of energy into and out of organisms.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- Plants, algae (including phytoplankton), and many microorganisms use the energy from light to make sugars (food) from carbon dioxide from the atmosphere and water through the process of photosynthesis, which also releases oxygen. These sugars can be used immediately or stored for growth or later use. (MS-LS-1.5)

Further Explanation: Emphasis is on tracing movement of matter and flow of energy.

Assessment Limit: Assessment does not include the biochemical mechanisms of photosynthesis.

MS-LS-1.6 Students who demonstrate understanding can:

Develop a conceptual model to describe how food is rearranged through chemical reactions forming new molecules that support growth and/or release energy as matter moves through an organism.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- Within individual organisms, food moves through a series of chemical reactions (cellular respiration) in which it is broken down and rearranged to form new molecules, to support growth, or to release energy. (MS-LS-1.6)

Further Explanation: Emphasis is on describing that molecules are broken apart and put back together and that in this process, energy is released and on understanding that the elements in the products are the same as the elements in the reactants.

Assessment Limit: Assessment does not include details of the chemical reactions for photosynthesis or respiration.

MS-LS-2.4 Students who demonstrate understanding can:

Develop a model to describe the flow of energy through the trophic levels of an ecosystem.

Supporting Content LS2.B: Cycling of Matter and Energy Transfer in Ecosystems

- Food webs can be broken down into multiple energy pyramids. Concepts should include the 10% rule of energy and biomass transfer between trophic levels and the environment. (MS-LS-2.4)

Further Explanation: Emphasis is on describing the transfer of mass and energy, beginning with producers, moving to primary and secondary consumers, and ending with decomposers.

Assessment Limit: Assessment does not include the use of chemical reactions to describe the processes.

MS-LS-2.5 Students who demonstrate understanding can:

Construct an argument supported by evidence that changes to physical or biological components of an ecosystem affect populations.

Supporting Content LS2.C: Ecosystem Dynamics, Functioning, and Resilience

- Ecosystems are dynamic in nature; their characteristics can vary over time. Disruptions to any physical or biological component of an ecosystem can lead to shifts in all its populations. (MS-LS-2.5)

Further Explanation: Emphasis is on recognizing patterns in data and making warranted inferences about changes in populations, and on evaluating empirical evidence supporting arguments about changes to ecosystems.

MS-LS-2.6 Students who demonstrate understanding can:

Design and evaluate solutions for maintaining biodiversity and ecosystem services.

Supporting Content LS2.C: Ecosystem Dynamics, Functioning, and Resilience

- Biodiversity describes the variety of species found in Earth's terrestrial and oceanic ecosystems. The completeness or integrity of an ecosystem's biodiversity is often used as a measure of its health. (MS-LS-2.6)

Supporting Content LS4.D: Biodiversity

- Changes in biodiversity can influence humans' resources, such as food, energy, and medicines, as well as ecosystem services that humans rely on—for example, water purification and recycling. (MS-LS-2.6)

Supporting Content ETS1.B: Developing Possible Solutions

- There are systematic processes for evaluating solutions with respect to how well they meet the criteria and constraints of a problem. (MS-LS-2.6)

Further Explanation: Examples of ecosystem services could include water purification, nutrient recycling, and prevention of soil erosion. Examples of design solution constraints could include scientific, economic, and social considerations.

MS-LS-4.4 Students who demonstrate understanding can:

Construct an explanation based on evidence that describes how genetic variations of traits in a population increase some individuals' probability of surviving and reproducing in a specific environment.

Supporting Content LS4.B: Natural Selection

- Natural selection leads to the predominance of certain traits in a population, and the suppression of others. (MS-LS-4.4)

Further Explanation: Emphasis is on using concepts of natural selection, including overproduction of offspring, passage of time, variation in a population, selection of favorable traits, and heritability of traits.

MS-LS-4.5 Students who demonstrate understanding can:

Obtain, evaluate, and communicate information about how technologies allow humans to influence the inheritance of desired traits in organisms.

Supporting Content LS4.B: Natural Selection

- In artificial selection, humans have the capacity to influence certain characteristics of organisms by selective breeding. One can choose desired parental traits determined by genes, which are then passed to offspring. (MS-LS-4.5)

Further Explanation: Emphasis is on identifying and communicating information from reliable sources about the influence of humans on genetic outcomes in artificial selection (such as genetic modification, animal husbandry, gene therapy), and on the influence these technologies have on society as well as the technologies leading to these scientific discoveries.

MS-LS-4.6 Students who demonstrate understanding can:

Use mathematical models to support explanations of how natural selection may lead to increases and decreases of specific traits in populations over time.

Supporting Content LS4.C: Adaptation

- Adaptation by natural selection acting over generations is one important process by which species change over time in response to changes in environmental conditions. Traits that support successful survival and reproduction in the new environment become more common; those that do not become less common. Thus, the distribution of traits in a population changes. (MS-LS-4.6)

Further Explanation: Emphasis is on using mathematical models, probability statements, and proportional reasoning to support explanations of trends in changes to populations over time. Examples could include Peppered Moth population changes before and after the industrial revolution.

Assessment Limit: Assessment does not include Hardy-Weinberg calculations.

MS-ESS-1.3 Students who demonstrate understanding can:

Analyze and interpret data to determine scale properties of objects in the solar system.

Supporting Content ESS1.B: Earth and the Solar System

- The solar system consists of the Sun and a collection of objects, including planets, their moons, and asteroids that are held in orbit around the Sun by its gravitational pull on them. (MS-ESS-1.2, MS-ESS-1.3)

Further Explanation: Emphasis is on the analysis of data from Earth-based instruments, space-based telescopes, and spacecraft to determine similarities and differences among solar system objects, such as relative size, distance, motions, and features. Examples of scale properties include the sizes of an object's layers (such as crust and atmosphere), surface features (such as volcanoes), and orbital radius. Examples of data include statistical information, drawings and photographs, and models.

Assessment Limit: Assessment does not include recalling facts about properties of the planets and other solar system bodies.

MS-ESS-1.4 Students who demonstrate understanding can:

Construct a scientific explanation based on evidence from rock strata for how the geologic time scale is used to analyze Earth's history.

Supporting Content ESS1.C: The History of Planet Earth

- The geologic time scale interpreted from rock strata provides a way to organize Earth's history. Analyses of rock strata and the fossil record provide only relative dates, not an absolute scale. (MS-ESS-1.4)

Further Explanation: Emphasis is on how analyses of rock formations and the fossils they contain are used to establish relative ages of major events in Earth's history. Examples of Earth's major events could range from being very recent (such as the last Ice Age) to very old (such as the formation of Earth or the earliest evidence of life). Examples can include the formation of mountain chains and ocean basins, the evolution or extinction of particular living organisms, or large volcanic eruptions.

Assessment Limit: Assessment does not include recalling the names of specific eons, eras, periods or epochs and events within them.

MS-ESS-2.3 Students who demonstrate understanding can:

Analyze and interpret data on the distribution of fossils and rocks, continental shapes, and seafloor structures to provide evidence of the past plate motions.

Supporting Content ESS1.C: The History of Planet Earth

- Tectonic processes continually generate new ocean sea floor at ridges and destroy old sea floor at trenches. (MS-ESS-2.3)

Supporting Content ESS2.B: Plate Tectonics and Large-Scale System Interactions

- Maps of ancient land and water patterns, based on investigations of rocks and fossils, make clear how Earth's plates have moved great distances, collided, and spread apart. (MS-ESS-2.3)

Further Explanation: Examples of data include similarities of rock and fossil types on different continents, the shapes of the continents (including continental shelves), and the locations of ocean structures (such as ridges, fracture zones, and trenches). Examples of concepts include continental drift and seafloor spreading.

Assessment Limit: Assessment of plate tectonics should be limited to large-scale system interactions. Paleomagnetic anomalies in oceanic and continental crust are not assessed.

MS-ESS-2.4 Students who demonstrate understanding can:

Develop a model to describe the cycling of water through Earth's systems driven by energy from the Sun and the force of gravity.

Supporting Content ESS2.C: The Roles of Water in Earth's Surface Processes

- Water continually cycles among land, ocean, and atmosphere via transpiration, evaporation, condensation, crystallization, percolation, and precipitation, as well as downhill flows on land. (MS-ESS-2.4)
- Global movements of water and its changes in form are propelled by sunlight and gravity. (MS-ESS-2.4)

Further Explanation: Emphasis is on the ways water changes its state as it moves through the multiple pathways of the hydrologic cycle. Examples of models can be conceptual or physical.

Assessment Limit: Assessment includes qualitative energy flows, not quantitative energy calculations.

MS-ESS-2.5 Students who demonstrate understanding can:

Collect data to provide evidence for how the motions and complex interactions of air masses results in changes in weather conditions.

Supporting Content ESS2.C: The Roles of Water in Earth's Surface Processes

- The complex patterns of the changes and the movement of water in the atmosphere, determined by winds, landforms, and ocean temperatures and currents, are major determinants of local weather patterns. (MS-ESS-2.5)

Supporting Content ESS2.D: Weather and Climate

- Because these patterns are so complex, weather can only be predicted using probability. (MS-ESS-2.5)

Further Explanation: Emphasis is on how air masses flow from regions of high pressure to low pressure, causing weather (defined by temperature, pressure, humidity, precipitation, and wind) at a fixed location to change over time, and how sudden changes in weather can result when different air masses collide. Emphasis is on how weather can be predicted within probabilistic ranges. Examples of data can be provided to students or obtained through laboratory experiments (such as with condensation and the use of barometers).

Assessment Limit: Assessment does not include recalling the names of cloud types or weather symbols used on weather maps or the reported diagrams from weather stations.

MS-ESS-2.6 Students who demonstrate understanding can:

Develop and use a model to describe how unequal heating and rotation of the Earth cause patterns of atmospheric and oceanic circulation that determine regional climates.

Supporting Content ESS2.C: The Roles of Water in Earth’s Surface Processes

- Variations in density due to variations in temperature and salinity drive a global pattern of interconnected ocean currents. (MS-ESS-2.6)

Supporting Content ESS2.D: Weather and Climate

- Weather and climate are influenced by interactions involving sunlight, the ocean, the atmosphere, ice, landforms, and living things. These interactions vary with latitude, altitude, and local and regional geography, all of which can affect oceanic and atmospheric flow patterns. (MS-ESS-2.6)
- The ocean exerts a major influence on weather and climate by absorbing energy from the Sun, releasing it over time, and globally redistributing it through ocean currents. (MS-ESS-2.6)

Further Explanation: Emphasis is on how patterns vary by latitude, altitude, and geographic land distribution. Emphasis of atmospheric circulation is on the sunlight-driven latitudinal banding, the Coriolis effect, and resulting prevailing winds; emphasis of ocean circulation is on the transfer of heat by the global ocean convection cycle, which is constrained by the Coriolis effect and the outlines of continents. Examples of models can be diagrams, maps and globes, or digital representations.

Assessment Limit: Assessment does not include the dynamics of the Coriolis effect, or recalling names and locations of specific biomes.

MS-ESS-3.4 Students who demonstrate understanding can:

Construct an argument based on evidence for how changes in human population and per-capita consumption of natural resources positively and negatively affect Earth's systems.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Technology and engineering can potentially help us best manage natural resources as populations increase. (MS-ESS-3.3, MS-ESS-3.4)

Further Explanation: Examples of evidence include grade-appropriate databases on human populations and the rates of consumption of food and natural resources (such as freshwater, mineral, and energy). Examples of effects can include changes made to the appearance, composition, and structure of Earth's systems as well as the rates at which they change.

MS-ESS-3.5 Students who demonstrate understanding can:

Ask questions to interpret evidence of the factors that cause climate variability throughout Earth's history.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Current scientific models indicate that human activities, such as the release of greenhouse gases from fossil fuel combustion, can contribute to the present-day measured rise in Earth's mean surface temperature. Natural activities, such as changes in incoming solar radiation, also contribute to changing global temperatures. (MS-ESS-3.5)

Further Explanation: Examples of factors include human activities (such as fossil fuel combustion and changes in land use) and natural processes (such as changes in incoming solar radiation and volcanic activity). Examples of evidence can include tables, graphs, and maps of global and regional temperatures; atmospheric levels of gases such as carbon dioxide and methane; and natural resource use.

HS-LS-1.4 Students who demonstrate understanding can:

Use a model to illustrate the role of cellular division (mitosis) and differentiation in producing and maintaining complex organisms.

Supporting Content LS1.B: Growth and Development of Organisms

- In multicellular organisms, individual cells grow and then divide via a process called mitosis, thereby allowing the organism to grow. The organism begins as a single cell (fertilized egg) that divides successively to produce many cells, with each parent cell passing identical genetic material (two variants of each chromosome pair) to both daughter cells. Cellular division and differentiation produce and maintain a complex organism, composed of systems of tissues and organs that work together to meet the needs of the whole organism. (HS-LS-1.4)

Assessment Limit: Assessment does not include specific gene control mechanisms.

HS-LS-1.5 Students who demonstrate understanding can:

Use a model to illustrate how photosynthesis transforms light energy into stored chemical energy.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- The process of photosynthesis converts light energy to stored chemical energy by converting carbon dioxide plus water into sugars plus released oxygen. (HS-LS-1.5)

Further Explanation: Emphasis is on illustrating inputs and outputs of matter and the transfer and transformation of energy in photosynthesis by plants and other photosynthesizing organisms. Examples of models could include diagrams, chemical equations, and conceptual models.

Assessment Limit: Assessment does not include specific biochemical steps.

HS-LS-1.6 Students who demonstrate understanding can:

Construct an explanation based on evidence for how carbon, hydrogen, and oxygen from sugar molecules may combine with other elements to form amino acids and/or other large carbon-based molecules.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- Sugar molecules contain carbon, hydrogen, and oxygen; their hydrocarbon backbones combined with nitrogen, sulfur and/or phosphorous are used to make monomers (amino acids) and other carbon-based molecules that can be assembled into larger macromolecules (such as proteins or DNA), used for example to form new cells. (HS-LS-1.6)
- As matter and energy flow through different organizational levels of living systems, chemical elements are recombined in different ways to form different products. (HS-LS-1.6, HS-LS-1.7)

Further Explanation: Emphasis is on using evidence from models and simulations to support explanations.

Assessment Limit: Assessment does not include the details of the specific chemical reactions or identification of macromolecules.

HS-LS-1.7 Students who demonstrate understanding can:

Use a model to illustrate that cellular respiration is a chemical process whereby the bonds of food molecules and oxygen molecules are broken and the bonds in new compounds are formed, resulting in a net transfer of energy.

Supporting Content LS1.C: Organization for Matter and Energy Flow in Organisms

- As matter and energy flow through different organizational levels of living systems, chemical elements are recombined in different ways to form different products. (HS-LS-1.6, HS-LS-1.7)
- As a result of these chemical reactions, energy is transferred from one system of interacting molecules to another. Cellular respiration is a chemical process in which the bonds of food molecules and oxygen molecules are broken, and new compounds are formed that can transport energy to cells. Cellular respiration also releases the energy needed to maintain body temperature despite ongoing energy transfer to the surrounding environment. (HS-LS-1.7)

Further Explanation: Emphasis is on the conceptual understanding of the inputs and outputs of the process of cellular respiration.

Assessment Limit: Assessment should not include identification of the steps or specific processes involved in cellular respiration.

HS-LS-2.3 Students who demonstrate understanding can:

Construct an explanation using mathematical representations to support claims for the flow of energy through trophic levels and the cycling of matter in an ecosystem.

Supporting Content LS2.B: Cycles of Matter and Energy Transfer in Ecosystems

- Photosynthesis and cellular respiration (including anaerobic processes) provide most of the energy for life processes. (HS-LS-2.3, HS-LS-2.4)
- Plants or algae form the lowest level of the food web. At each link upward in a food web, only a small fraction of the matter consumed at the lower level is transferred upward, to produce growth and release energy in cellular respiration at the higher level. Given this inefficiency, there are generally fewer organisms at higher levels of a food web. Some matter reacts to release energy for life functions, some matter is stored in newly made structures, and much is discarded. The chemical elements that make up the molecules of organisms pass through food webs and into and out of the atmosphere and soil, and they are combined and recombined in different ways. At each link in an ecosystem, matter and energy are conserved. (HS-LS-2.3)

Further Explanation: Emphasis is on using a mathematical model of stored energy in biomass to describe the transfer of energy from one trophic level to another and that matter and energy are conserved as matter cycles and energy flows through ecosystems. Emphasis is on atoms and molecules such as carbon, oxygen, hydrogen and nitrogen being conserved as they move through an ecosystem.

Assessment Limit: Assessment is limited to proportional reasoning to describe the cycling of matter and flow of energy.

HS-LS-2.4 Students who demonstrate understanding can:

Develop a model to illustrate the role of photosynthesis and cellular respiration in the cycling of carbon among the biosphere, atmosphere, hydrosphere, and geosphere.

Supporting Content LS2.B: Cycles of Matter and Energy Transfer in Ecosystems

- Photosynthesis and cellular respiration are important components of the carbon cycle, in which carbon is exchanged among the biosphere, atmosphere, oceans, and geosphere through chemical, physical, geological, and biological processes. (HS-LS-2.4)
- Photosynthesis and cellular respiration (including anaerobic processes) provide most of the energy for life processes. (HS-LS-2.3, HS-LS-2.4)

Further Explanation: Examples of models could include simulations and mathematical models.

Assessment Limit: Assessment does not include the specific chemical steps of photosynthesis and respiration.

HS-LS-2.5 Students who demonstrate understanding can:

Evaluate the claims, evidence, and reasoning that changing the conditions of a static ecosystem may result in a new ecosystem.

Supporting Content LS2.C: Ecosystem Dynamics, Functioning, and Resilience

- A complex set of interactions within an ecosystem can keep its numbers and types of organisms relatively constant over long periods of time under stable conditions. If a modest biological or physical disturbance to an ecosystem occurs, it may return to its more or less original status (i.e., the ecosystem is resilient), as opposed to becoming a very different ecosystem. Extreme fluctuations in conditions or the size of any population, however, can challenge the functioning of ecosystems in terms of resources and habitat availability. (HS-LS-2.2, HS-LS-2.5)

Further Explanation: Examples of changes in ecosystem conditions could include modest biological or physical changes, such as a seasonal flood, and extreme changes, such as volcanic eruption or sea level rise.

HS-LS-2.6 Students who demonstrate understanding can:

Design, evaluate, and/or refine practices used to manage a natural resource based on direct and indirect influences of human activities on biodiversity and ecosystem health.

Supporting Content LS2.C: Ecosystem Dynamics, Functioning, and Resilience

- Changes in the environment, including habitat destruction, pollution, introduction of invasive species, overexploitation, and climate variability, can disrupt an ecosystem and threaten the survival of some species. (HS-LS-2.6)

Supporting Content LS4.C: Adaptation

- Changes in the physical environment have contributed to the expansion of some species, the emergence of new distinct species as populations diverge under different conditions, and the decline or possible extinction of some species. (HS-LS-2.6, HS-LS-4.5)

Supporting Content LS4.D: Biodiversity

- Biodiversity is increased by the formation of new species (speciation) and decreased by the loss of species (extinction). (HS-LS-2.6)
- Sustaining ecosystem health and biodiversity is essential to support and enhance life on Earth. Sustaining biodiversity also aids humanity by preserving landscapes of recreational, cultural, or inspirational value. Humans depend on the living world for the resources and other benefits provided by biodiversity. Effects on biodiversity can be mitigated through actions such as habitat conservation, reclamation practices, wildlife management, and invasive species control. Understanding the effects of population growth, wildfire, pollution, and climate variability on changes in biodiversity could help maintain the integrity of biological systems. (HS-LS-2.6)

Supporting Content ESS3.A: Natural Resources

- Resource availability has guided the development of human society. (HS-LS-2.6, HS-ESS-3.1)

Supporting Content ETS1.B: Developing Possible Solutions

- When evaluating solutions, it is important to take into account a range of constraints, including cost, safety, reliability, and aesthetics, and to consider social, environmental, and cultural impacts. (HS-LS-2.6)

Further Explanation: Emphasis is on how natural resources such as forests, waterways, and land are managed in ways that minimize harm to biodiversity and ecosystem health and activities that can improve and or maintain existing health of ecosystems.

HS-LS-2.7 Students who demonstrate understanding can:

Evaluate the evidence for the role of group behavior on individual and species' ability to survive and reproduce.

Supporting Content LS2.D: Social Interactions and Group Behavior

- Group behavior has evolved because membership can increase the chances of survival for individuals and their genetic relatives. (HS-LS-2.7)

Further Explanation: Emphasis is on: (1) distinguishing between group and individual behavior, (2) identifying evidence supporting the outcomes of group behavior, and (3) developing logical and reasonable arguments based on evidence. Examples of group behaviors could include flocking, schooling, herding, and cooperative behaviors such as hunting, migrating, and swarming.

HS-LS-4 – Biological Adaptation: Unity and Diversity

HS-LS-4.1 Students who demonstrate understanding can:

Communicate scientific information that common ancestry and biological evolution are supported by multiple lines of empirical evidence.

Supporting Content LS4.A: Evidence of Common Ancestry and Diversity

- Genetic information, like the fossil record, provides evidence of evolution. DNA sequences vary among species, but there are many overlaps; the ongoing branching that produces multiple lines of descent can be inferred by comparing the DNA sequences of different organisms. Such information can be derived from the similarities and differences in amino acid sequences and from anatomical and embryological evidence. (HS-LS-4.1)

Further Explanation: Emphasis is on a conceptual understanding of the role each line of evidence has relating to common ancestry and biological evolution. Examples of evidence could include similarities in DNA sequences, anatomical structures, and order of appearance of structures in embryological development.

HS-LS-4.2 Students who demonstrate understanding can:

Construct an explanation based on evidence that the process of evolution, through the mechanism of natural selection, primarily results from four factors: (1) the potential for a species to increase in number, (2) the heritable genetic variation of individuals in a species due to mutation and sexual reproduction, (3) competition for limited resources, and (4) the proliferation of those organisms that are better able to survive and reproduce in the environment.

Supporting Content LS4.B: Natural Selection

- Natural selection occurs only if there is both (1) variation in the genetic information between organisms in a population and (2) variation in the expression of that genetic information—that is, trait variation—that leads to differences in performance among individuals. (HS-LS-4.2, HS-LS-4.3)

Supporting Content LS4.C: Adaptation

- Evolution is a consequence of the interaction of four factors of natural selection: (1) the potential for a species to increase in number, (2) the genetic variation of individuals in a species due to mutation and sexual reproduction, (3) competition for an environment's limited supply of the resources that individuals need in order to survive and reproduce, and (4) the ensuing proliferation of those organisms that are better able to survive and reproduce in that environment. (HS-LS-4.2)

Further Explanation: Emphasis is on using evidence to explain the influence each of the four factors has on the number of organisms, behaviors, morphology, or physiology in terms of ability to compete for limited resources and subsequent survival of individuals and adaptation of species. Examples of evidence could include mathematical models such as simple distribution graphs and proportional reasoning.

Assessment Limit: Assessment does not include other mechanisms of evolution, such as genetic drift, gene flow through migration, and co-evolution.

HS-LS-4.3 Students who demonstrate understanding can:

Apply concepts of probability and statistical analysis to support explanations that organisms with an advantageous heritable trait tend to increase in proportion to organisms lacking this trait.

Supporting Content LS4.B: Natural Selection

- Natural selection occurs only if there is both (1) variation in the genetic information between organisms in a population and (2) variation in the expression of that genetic information—that is, trait variation—that leads to differences in performance among individuals. (HS-LS-4.2, HS-LS-4.3)
- The traits that positively affect survival are more likely to be reproduced, and thus are more common in the population. (HS-LS-4.3)

Supporting Content LS4.C: Adaptation

- Natural selection leads to adaptation, that is, to a population dominated by organisms that are anatomically, behaviorally, and physiologically well suited to survive and reproduce in a specific environment. That is, the differential survival and reproduction of organisms in a population that have an advantageous heritable trait leads to an increase in the proportion of individuals in future generations that have the trait and to a decrease in the proportion of individuals that do not. (HS-LS-4.3, HS-LS-4.4)
- Adaptation also means that the distribution of traits in a population can change when conditions change. (HS-LS-4.3)

Further Explanation: Emphasis is on analyzing shifts in numerical distribution of traits and using these shifts as evidence to support explanations.

Assessment Limit: Assessment is limited to basic statistical and graphical analysis. Assessment does not include allele frequency calculations.

HS-LS-4.4 Students who demonstrate understanding can:

Construct an explanation based on evidence for how natural selection leads to adaptation of populations.

Supporting Content LS4.C: Adaptation

- Natural selection leads to adaptation, that is, to a population dominated by organisms that are anatomically, behaviorally, and physiologically well suited to survive and reproduce in a specific environment. The differential survival and reproduction of organisms in a population that have an advantageous heritable trait leads to an increase in the proportion of individuals in future generations that have the trait and to a decrease in the proportion of individuals that do not. (HS-LS-4.3, HS-LS-4.4)

Further Explanation: Emphasis is on using data to provide evidence for how specific biotic and abiotic differences in ecosystems (such as ranges of seasonal temperature, long-term climate change, acidity, light, geographic barriers, or evolution of other organisms) contribute to a change in gene frequency over time, leading to adaptation of populations.

HS-LS-4.5 Students who demonstrate understanding can:

Evaluate models that demonstrate how changes in an environment may result in the evolution of a population of a given species; the emergence of new species over generations; or the extinction of other species due to the processes of genetic drift, gene flow, mutation, and natural selection.

Supporting Content LS4.C: Adaptation

- Changes in the physical environment have contributed to the expansion of some species, the emergence of new distinct species as populations diverge under different conditions, and the decline or possible extinction of some species. (HS-LS-2.6, HS-LS-4.5)
- Species become extinct because they can no longer survive and reproduce in their altered environment. If members cannot adjust to change that is too fast or drastic, the opportunity for the species' evolution is lost. (HS-LS-4.5)

Further Explanation: Emphasis is on determining cause and effect relationships for how changes to the environment such as drought, flood, fire, deforestation, overfishing, application of fertilizers and pesticides, and the rate of change of the environment affect the distribution or disappearance of traits in species.

HS-PSC-1.3 Students who demonstrate understanding can:

Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrostatic forces between particles.

Supporting Content PS1.A: Structure and Properties of Matter

- The structure and interactions of matter at the bulk scale are determined by electrical forces within and between atoms. (HS-PSC-1.3, HS-PSC-1.5)

Supporting Content PS2.B: Types of Interactions

- Attraction and repulsion between electric charges at the atomic scale explain the structure, properties (physical and chemical), and transformations of matter, as well as the contact forces between material objects. (HS-PSC-1.3, HS-PSC-1.5, *HS-PSP-1.6*)

Further Explanation: Emphasis is on understanding the strengths of forces between particles.

Examples of particles could include ions, atoms, molecules, and networked materials (such as graphite). Examples of bulk properties of substances could include the melting point and boiling point, vapor pressure, and surface tension.

Assessment Limit: Assessment does not include naming specific intermolecular forces (such as dipole-dipole). Assessment will be limited to quantitative calculations of melting and boiling points only.

HS-PSC-1.4 Students who demonstrate understanding can:

Develop models to illustrate the changes in the composition of the nucleus of the atom and the energy released during the processes of fission, fusion, and the various modes of radioactive decay.

Supporting Content PS1.C: Nuclear Processes

- Nuclear processes, including fusion, fission, and radioactive decays of unstable nuclei, involve release or absorption of energy. The total number of neutrons plus protons does not change in any nuclear process. (HS-PSC-1.4)

Further Explanation: Emphasis is on simple qualitative models, such as pictures or diagrams, and on the scale of energy released in nuclear processes relative to other kinds of transformations.

Assessment Limit: Assessment does not include quantitative calculation of energy released. Assessment is limited to alpha, beta, and gamma modes of radioactive decay.

HS-PSC-1.5 Students who demonstrate understanding can:

Communicate scientific and technical information about why the molecular-level structure is important in the functioning of designed materials.

Supporting Content PS1.A: Structure and Properties of Matter

- The structure and interactions of matter at the bulk scale are determined by electrical forces within and between atoms. (HS-PSC-1.3, HS-PSC-1.5)

Supporting Content PS2.B: Types of Interactions

- Attraction and repulsion between electric charges at the atomic scale explain the structure, properties (physical and chemical), and transformations of matter, as well as the contact forces between material objects. (HS-PSC-1.3, HS-PSC-1.5, *HS-PSP-1.6*)

Further Explanation: Emphasis is on the attractive and repulsive forces that determine the functioning of the material. Examples could include why electrically conductive materials are often made of metal, flexible but durable materials are made up of long chained molecules, and pharmaceuticals are designed to interact with specific receptors.

Assessment Limit: Assessment is limited to provided molecular structures of specific designed materials. For questions involving polar vs. nonpolar bonds, item distractors containing ionic bonds may not be used. Electronegativity differences of < 0.5 should be used for nonpolar covalent bonds. Electronegativity differences of $0.5 - 1.7$ should be used for polar covalent bonds.

HS-PSC-2 – Chemical Reactions

HS-PSC-2.1 Students who demonstrate understanding can:

Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.

Supporting Content PS1.A: Structure and Properties of Matter

- The periodic table orders elements horizontally by the number of protons in the atom's nucleus and places those with similar physical and chemical properties in columns. The repeating patterns of this table reflect patterns of outer electron states. (HS-PSC-2.1)

Supporting Content PS1.B: Chemical Reactions

- The fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions. (HS-PSC-2.1, HS-PSC-2.4)

Further Explanation: Examples of chemical reactions could include the reaction of sodium and chlorine, of carbon and oxygen, or of carbon and hydrogen.

Assessment Limit: Assessment is limited to synthesis, decomposition, single replacement/displacement, double replacement/displacement—including neutralization—and combustion reactions. Predict the products of double replacement, single replacement, and combustion reactions only. Assessment excludes writing formulas or names of acids and hydrocarbons.

HS-PSC-2.2 Students who demonstrate understanding can:

Develop a model to illustrate that the energy transferred during an exothermic or endothermic chemical reaction is based on the bond energy difference between bonds broken (absorption of energy) and bonds formed (release of energy).

Supporting Content PS1.A: Structure and Properties of Matter

- A stable molecule has less energy than the same set of atoms separated; one must provide at least this energy in order to take the molecule apart. (HS-PSC-2.2)

Supporting Content PS1.B: Chemical Reactions

- Chemical processes, their rates, and whether or not energy is stored or released can be understood in terms of the collisions of molecules and the rearrangements of atoms into new molecules, with consequent changes in the sum of all bond energies in the set of molecules that are matched by changes in kinetic energy. (HS-PSC-2.2, HS-PSC-2.3)

Further Explanation: Emphasis is on the idea that a chemical reaction is a system that affects the energy change. Examples of models could include molecular-level drawings and diagrams of reactions, graphs showing the relative energies of reactants and products, and representations showing energy is conserved.

Assessment Limit: Assessment does not include calculating the total bond energy changes during a chemical reaction from the bond energies of reactants and products.

HS-PSC-2.3 Students who demonstrate understanding can:

Apply scientific principles and evidence to provide an explanation about the effects of changing the temperature or concentration of the reacting particles on the rate at which a reaction occurs.

Supporting Content PS1.B: Chemical Reactions

- Chemical processes, their rates, and whether or not energy is stored or released can be understood in terms of the collisions of molecules and the rearrangements of atoms into new molecules, with consequent changes in the sum of all bond energies in the set of molecules that are matched by changes in kinetic energy. (HS-PSC-2.2, HS-PSC-2.3)

Further Explanation: Emphasis is on student reasoning that focuses on the number and energy of collisions between molecules.

Assessment Limit: Assessment is limited to simple reactions in which there are only two reactants; evidence from temperature, concentration, and rate data; and qualitative relationships between rate and temperature.

HS-PSC-2.4 Students who demonstrate understanding can:

Use mathematical representations to support the claim that the number and type of atoms, and therefore mass, are conserved during a chemical reaction.

Supporting Content PS1.B: Chemical Reactions

- The fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions. (HS-PSC-2.1, HS-PSC-2.4)

Further Explanation: Emphasis is on using mathematical ideas to communicate the proportional relationships between masses of atoms in the reactants and the products, and the translation of these relationships to the macroscopic scale using the mole as the conversion from the atomic to the macroscopic scale. Emphasis is on assessing students' use of mathematical thinking and not on memorization and rote application of problem-solving techniques.

Assessment Limit: Conversion problems will be one to two steps (e.g., grams to moles to atoms/molecules). Compounds and formulas should be provided in the stem of the question. Students should be given molecular masses in problems involving gram to other unit conversions. Molar mass calculations should not be combined with conversion problems. All volumes must be at standard temperature and pressure (STP). A balanced equation and molar masses should be included in the item. Calculations may include grams/moles/volume of reactant to grams/moles/volume of product.

HS-PSC-3.3 Students who demonstrate understanding can:

Develop and use models to illustrate that energy at the macroscopic scale can be accounted for as a combination of energy associated with the motions of particles (objects) and energy associated with the relative positions of particles (objects).

Supporting Content PS3.A: Definitions of Energy

- Energy is a quantitative property of a system that depends on the motion and interactions of matter and radiation within that system. That there is a single quantity called energy is due to the fact that a system's total energy is conserved, even as, within the system, energy is continually transferred from one object to another and between its various possible forms. (HS-PSC-3.2, HS-PSC-3.3)
- At the macroscopic scale, energy manifests itself in multiple ways, such as in motion, sound, light, and thermal energy. (HS-PSC-3.3, HS-PSC-3.4)
- These relationships are better understood at the microscopic scale, at which all of the different manifestations of energy can be modeled as a combination of energy associated with the motion of particles and energy associated with the configuration (relative position of the particles). In some cases, the relative position energy can be thought of as stored in fields (which mediate interactions between particles). This last concept includes radiation, a phenomenon in which energy stored in fields moves across space. (HS-PSC-3.3)

Further Explanation: Examples of phenomena at the macroscopic scale could include the conversion of kinetic energy to thermal energy, the energy stored due to position of an object above the earth, and the energy stored between two electrically charged plates. Examples of models could include diagrams, drawings, descriptions, and computer simulations.

HS-PSC-3.4* Students who demonstrate understanding can:

Design, build, and refine a device that works within given constraints to convert one form of energy into another form of energy. ---OPTIONAL

Supporting Content PS3.A: Definitions of Energy

- At the macroscopic scale, energy manifests itself in multiple ways, such as in motion, sound, light, and thermal energy. (HS-PSC-3.3, HS-PSC-3.4)

Supporting Content PS3.D: Energy in Chemical Processes

- Although energy cannot be destroyed, it can be converted to less useful forms—for example, to thermal energy in the surroundings. (HS-PSC-3.4, HS-PSC-3.5)

Further Explanation: Emphasis is on both qualitative and quantitative evaluations of devices.

Examples of devices could include calorimeters, heat and cold packs, solar cells, solar ovens, and electrochemical cells. Examples of constraints could include use of renewable energy forms and efficiency.

Assessment Limit: Assessment for quantitative evaluations is limited to total output for a given input. Assessment is limited to devices constructed with materials provided to students.

HS-PSC-3.5 Students who demonstrate understanding can:

Plan and conduct an investigation to provide evidence that the transfer of thermal energy when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (second law of thermodynamics).

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- Energy cannot be created or destroyed, but it can be transported from one place to another and transferred between systems. (HS-PSC-3.2, HS-PSC-3.5)
- Uncontrolled systems always evolve toward more stable states—that is, toward a more uniform energy distribution (e.g., water flows downhill, objects hotter than their surrounding environment cool down). (HS-PSC-3.5)

Supporting Content PS3.D: Energy in Chemical Processes

- Although energy cannot be destroyed, it can be converted to less useful forms—for example, to thermal energy in the surroundings. (HS-PSC-3.4, HS-PSC-3.5)

Further Explanation: Emphasis is on analyzing data from student investigations and using mathematical thinking to describe the energy changes both quantitatively and conceptually (endothermic/exothermic). Examples of investigations could include mixing liquids at different initial temperatures or adding objects at different temperatures to water.

Assessment Limit: For items involving specific heat, provide the equation $Q = mC_p\Delta T$ and specific heats. Include the melting and boiling points of water. Limit calculations to changes that do not involve a change of state. Perform gram to mole and mole to ΔH calculations. Use joules as a unit of measure, as opposed to calories.

HS-PSP-1.3 Students who demonstrate understanding can:

Apply scientific and engineering ideas to design, evaluate, and refine a device that minimizes the force on a macroscopic object during a collision.

Supporting Content PS2.A: Forces and Motion

- If a system interacts with objects outside itself, the total momentum of the system can change; however, any such change is balanced by changes in the momentum of objects outside the system. (HS-PSP-1.2, HS-PSP-1.3)

Supporting Content ETS1.A: Defining and Delimiting an Engineering Problem

- Criteria and constraints also include satisfying any requirements set by society, such as taking issues of risk mitigation into account, and they should be quantified to the extent possible and stated in such a way that one can tell if a given design meets them. (HS-PSP-1.3)

Supporting Content ETS1.C: Optimizing the Design Solution

- Criteria may need to be broken down into simpler ones that can be approached systematically, and decisions about the priority of certain criteria over others (trade-offs) may be needed. (HS-PSP-1.3)

Further Explanation: Examples of evaluation and refinement could include determining the success of the device at protecting an object from damage and modifying the design to improve it. Examples of a device could include a football helmet or a parachute.

Assessment Limit: Assessment is limited to qualitative evaluations and/or algebraic manipulations.

HS-PSP-1.4 Students who demonstrate understanding can:

Use mathematical representations of Newton’s Law of Gravitation and Coulomb’s Law to describe and predict the gravitational and electrostatic forces between objects.

Supporting Content PS2.B: Types of Interactions

- Newton’s law of universal gravitation and Coulomb’s law provide the mathematical models to describe and predict the effects of gravitational and electrostatic forces between distant objects. (HS-PSP-1.4)
- Forces at a distance are explained by fields (gravitational, electric, and magnetic) permeating space that can transfer energy through space. Magnets or electric currents cause magnetic fields; electric charges or changing magnetic fields cause electric fields. (HS-PSP-1.4, HS-PSP-1.5)

Further Explanation: Emphasis is on both quantitative and conceptual descriptions of gravitational and electric fields.

Assessment Limit: Assessment is limited to systems with two objects. Base equations will be provided.

HS-PSP-1.5 Students who demonstrate understanding can:

Plan and conduct an investigation to provide evidence that an electric current can produce a magnetic field and that a changing magnetic field can produce an electric current.

Supporting Content PS2.B: Types of Interactions

- Forces at a distance are explained by fields (gravitational, electric, and magnetic) permeating space that can transfer energy through space. Magnets or electric currents cause magnetic fields; electric charges or changing magnetic fields cause electric fields. (HS-PSP-1.4, HS-PSP-1.5)

Supporting Content PS3.A: Definitions of Energy

- “Electrical energy” may mean energy stored in a battery or energy transmitted by electric currents. (HS-PSP-1.5)

Assessment Limit: Assessment is limited to designing and conducting investigations with provided materials and tools.

HS-PSP-1.6 Students who demonstrate understanding can:

Communicate scientific and technical information about why the molecular-level structure is important in the functioning of designed materials.

Supporting Content PS1.A: Structure and Properties of Matter

- The structure and interactions of matter at the bulk scale are determined by electrical forces within and between atoms. (HS-PSP-1.6)

Supporting Content PS2.B: Types of Interactions

- Attraction and repulsion between electric charges at the atomic scale explain the structure, properties, and transformations of matter, as well as the contact forces between material objects. (HS-PSP-1.6, HS-PSC-1.3, HS-PSC-1.5)

Further Explanation: Emphasis is on the attractive and repulsive forces that determine the functioning of the material. Examples could include why electrically conductive materials are often made of metal, flexible but durable materials are made up of long chained molecules, and pharmaceuticals are designed to interact with specific receptors.

Assessment Limit: Assessment is limited to provided molecular structures of specific designed materials.

HS-PSP-2.3 Students who demonstrate understanding can:

Design, build, and refine a device that works within given constraints to convert one form of energy into another form of energy.

Supporting Content PS3.A: Definitions of Energy

- At the macroscopic scale, energy manifests itself in multiple ways, such as in motion, sound, light, and thermal energy. (HS-PSP-2.2, HS-PSP-2.3)

Supporting Content PS3.D: Energy in Chemical Processes

- Although energy cannot be destroyed, it can be converted to less useful forms—for example, to thermal energy in the surrounding environment. (HS-PSP-2.3, HS-PSP-2.4)

Supporting Content ETS1.A: Defining and Delimiting an Engineering Problem

- Criteria and constraints also include satisfying any requirements set by society, such as taking issues of risk mitigation into account, and they should be quantified to the extent possible and stated in such a way that one can tell if a given design meets them. (HS-PSP-2.3)

Further Explanation: Emphasis is on both qualitative and quantitative evaluations of devices.

Examples of devices could include Rube Goldberg devices, wind turbines, solar cells, solar ovens, and generators. Examples of constraints could include use of multiple energy forms and evaluations of efficiency.

Assessment Limit: Assessment for quantitative evaluations is limited to total output for a given input. Assessment is limited to examples of devices provided to students.

HS-PSP-2.4 Students who demonstrate understanding can:

Plan and conduct an investigation to provide evidence that the transfer of thermal energy when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (second law of thermodynamics).

Supporting Content PS3.B: Conservation of Energy and Energy Transfer

- Energy cannot be created or destroyed, but it can be transported from one place to another and transferred between systems. (HS-PSP-2.1, HS-PSP-2.4)
- Uncontrolled systems always evolve toward more stable states—that is, toward a more uniform energy distribution (e.g., water flows downhill, objects hotter than their surrounding environment cool down). (HS-PSP-2.4)

Supporting Content PS3.D: Energy in Chemical Processes

- Although energy cannot be destroyed, it can be converted to less useful forms—for example, to thermal energy in the surrounding environment. (HS-PSP-2.3, HS-PSP-2.4)

Further Explanation: Emphasis is on analyzing data from student investigations and using mathematical thinking to describe the energy changes both quantitatively and conceptually.

Examples of investigations could include mixing liquids at different initial temperatures or adding objects at different temperatures to water.

Assessment Limit: Assessment is limited to examples of closed system investigations.

HS-PSP-2.5 Students who demonstrate understanding can:

Develop and use a model of two objects interacting through electric or magnetic fields to illustrate the forces between objects and the changes in energy of the objects due to the interaction.

Supporting Content PS3.C: Relationship Between Energy and Forces

- When two objects interacting through a field change relative position, the energy stored in the field is changed. (HS-PSP-2.5)

Further Explanation: Examples of models could include drawings, diagrams, and texts, such as drawings of what happens when two charges of opposite polarity are near each other.

Assessment Limit: Assessment is limited to systems containing two objects.

HS-PSP-3.4 Students who demonstrate understanding can:

Evaluate the validity and reliability of claims in published materials of the effects that different frequencies of electromagnetic radiation have when absorbed by matter.

Supporting Content PS4.B: Electromagnetic Radiation

- When light or longer wavelength electromagnetic radiation is absorbed in matter, it is generally converted into thermal energy (heat). Shorter wavelength electromagnetic radiation (ultraviolet, X-rays, gamma rays) can ionize atoms and cause damage to living cells. (HS-PSP-3.4)

Further Explanation: Emphasis is on the idea that photons associated with different frequencies of light have different energies, and the damage to living tissue from electromagnetic radiation depends on the energy of the radiation. Examples of published materials could include trade books, magazines, web resources, videos, and other passages that may reflect bias.

Assessment Limit: Assessment is limited to qualitative descriptions.

HS-PSP-3.5 Students who demonstrate understanding can:

Communicate technical information about how some technological devices use the principles of wave behavior and wave interactions with matter to transmit and capture information and energy.

Supporting Content PS3.D: Energy in Chemical Processes

- Solar cells are human-made devices that likewise capture the Sun’s energy and produce electrical energy. (HS-PSP-3.5)

Supporting Content PS4.A: Wave Properties

- Information can be digitized (e.g., a picture stored as the values of an array of pixels); in this form, it can be stored reliably in computer memory and sent over long distances as a series of wave pulses. (HS-PSP-3.2, HS-PSP-3.5)

Supporting Content PS4.B: Electromagnetic Radiation

- Photoelectric materials emit electrons when they absorb light of a high enough frequency. (HS-PSP-3.5)

Supporting Content PS4.C: Information Technologies and Instrumentation

- Multiple technologies based on the understanding of waves and their interactions with matter are part of everyday experiences in the modern world (e.g., medical imaging, communications, scanners) and in scientific research. They are essential tools for producing, transmitting, and capturing signals and for storing and interpreting the information contained in them. (HS-PSP-3.5)

Further Explanation: Examples could include solar cells capturing light and converting it to electricity, medical imaging, and communications technology.

Assessment Limit: Assessments are limited to qualitative information. Assessments do not include band theory.

HS-ESS-1.3 Students who demonstrate understanding can:

Communicate scientific ideas about the way stars, over their life cycle, transform elements.

Supporting Content ESS1.A: The Universe and Its Stars

- The study of stars' light spectra and brightness is used to identify compositional elements of stars, their movements, and their distances from Earth. (HS-ESS-1.2, HS-ESS-1.3)
- Origin theories are supported by evidence such as observations of distant galaxies receding from our own, of the measured composition of stars and non-stellar gases, and of the maps of spectra of the primordial radiation (cosmic microwave background) that still fills the universe. Other than the hydrogen and helium formed at the time of the event, nuclear fusion within stars produces all atomic nuclei lighter than and including iron, and the process releases electromagnetic energy. Heavier elements are produced when certain massive stars achieve a supernova stage and explode. (HS-ESS-1.2, HS-ESS-1.3)

Further Explanation: Emphasis is on the way nucleosynthesis, and therefore the different elements created, varies as a function of the mass of a star and the stage of its lifetime.

Assessment Limit: Details of the many different nucleosynthesis pathways for stars of different masses are not assessed.

HS-ESS-1.4 Students who demonstrate understanding can:

Use mathematical or computational representations to predict the motion of orbiting objects in the solar system.

Supporting Content ESS1.B: Earth and the Solar System

- Kepler's laws describe common features of the motions of orbiting objects, including their elliptical paths around the Sun. Orbits may change due to the gravitational effects from, or collisions with, other objects in the solar system. (HS-ESS-1.4)

Further Explanation: Emphasis is on Newtonian gravitational laws governing orbital motions, which apply to human-made satellites as well as planets and moons.

Assessment Limit: Mathematical representations for the gravitational attraction of bodies and Kepler's Laws of orbital motions should not deal with more than two bodies, nor involve calculus.

HS-ESS-1.5 Students who demonstrate understanding can:

Evaluate evidence of the past and current movements of continental and oceanic crust and the theory of plate tectonics to explain the ages of crustal rocks.

Supporting Content ESS1.C: The History of Planet Earth

- Continental rocks are generally much older than the rocks of the ocean floor. (HS-ESS-1.5)

Supporting Content ESS2.B: Plate Tectonics and Large-Scale System Interactions

- Plate tectonics is the unifying theory that explains the past and current movements of the rocks at Earth's surface and provides a framework for understanding its geologic history. (HS-ESS-1.5)

Supporting Content PS1.C: Nuclear Processes

- Spontaneous radioactive decay follows a characteristic exponential decay law. Nuclear lifetimes allow radiometric dating to be used to determine the ages of rocks and other materials. (HS-ESS-1.5, HS-ESS-1.6)

Further Explanation: Emphasis is on the ability of plate tectonics to explain the ages of crustal rocks. Examples include evidence of the ages of oceanic crust increasing with distance from mid-ocean ridges (a result of plate spreading) and the ages of North American continental crust decreasing with distance away from a central ancient core (a result of past plate interactions).

HS-ESS-1.6 Students who demonstrate understanding can:

Apply scientific reasoning and evidence from ancient Earth materials, meteorites, and other planetary surfaces to construct an account of Earth's formation and early history.

Supporting Content ESS1.C: The History of Planet Earth

- Although active geologic processes, such as plate tectonics and erosion, have destroyed or altered most of the very early rock record on Earth, other objects in the solar system, such as lunar rocks, asteroids, and meteorites, have changed little over billions of years. Studying these objects can provide information about Earth's formation and early history. (HS-ESS-1.6)

Supporting Content PS1.C: Nuclear Processes

- Spontaneous radioactive decay follows a characteristic exponential decay law. Nuclear lifetimes allow radiometric dating to be used to determine the ages of rocks and other materials. (HS-ESS-1.5, HS-ESS-1.6)

Further Explanation: Emphasis is on using available evidence within the solar system to reconstruct the early history of Earth, which formed along with the rest of the solar system. Examples of evidence include the absolute ages of ancient materials (obtained by radiometric dating of meteorites, Moon rocks, and Earth's oldest minerals); the sizes and compositions of solar system objects; and the impact cratering record of planetary surfaces.

HS-ESS-2.3 Students who demonstrate understanding can:

Develop a model based on evidence of Earth's interior to describe the cycling of matter by thermal convection.

Supporting Content ESS2.A: Earth Material and Systems

- Evidence from deep probes and seismic waves, reconstructions of historical changes in Earth's surface and its magnetic field, and an understanding of physical and chemical processes lead to a model of Earth with a hot but solid inner core, a liquid outer core, a solid mantle, and crust. Motions of the mantle and its plates occur primarily through thermal convection, which involves the cycling of matter due to the outward flow of energy from Earth's interior and gravitational movement of denser materials toward the interior. (HS-ESS-2.3)

Supporting Content ESS2.B: Plate Tectonics and Large-Scale System Interactions

- The radioactive decay of unstable isotopes continually generates new energy within Earth's crust and mantle, providing the primary source of the heat that drives mantle convection. Plate tectonics can be viewed as the surface expression of mantle convection. (HS-ESS-2.3)

Supporting Content PS4.A: Wave Properties

- Geologists use seismic waves and their reflection at interfaces between layers to probe structures deep in the planet. (HS-ESS-2.3)

Further Explanation: Emphasis is on both a one-dimensional model of Earth, with radial layers determined by density, and a three-dimensional model, which is controlled by mantle convection and the resulting plate tectonics. Examples of evidence include maps of Earth's three-dimensional structure obtained from seismic waves, records of the rate of change of Earth's magnetic field (as constraints on convection in the outer core), and identification of the composition of Earth's layers from high-pressure laboratory experiments.

HS-ESS-2.4 Students who demonstrate understanding can:

Use a model to describe how variations in the flow of energy into and out of Earth's systems result in variations in climate.

Supporting Content ESS1.B: Earth and the Solar System

- Cyclical changes in the shape of Earth's orbit around the Sun, together with changes in the tilt of the planet's axis of rotation, both occurring over hundreds of thousands of years, have altered the intensity and distribution of sunlight falling on the earth. These phenomena cause a cycle of ice ages and other gradual climate variations. (HS-ESS-2.4)

Supporting Content ESS2.A: Earth Material and Systems

- The geological record shows that changes to global and regional climate can be caused by interactions among changes in the Sun's energy output or Earth's orbit, tectonic events, ocean circulation, volcanic activity, glaciers, vegetation, and human activities. These changes can occur on a variety of time scales from sudden (e.g., volcanic ash clouds) to intermediate (e.g., ice ages) to very long-term tectonic cycles. (HS-ESS-2.4)

Supporting Content ESS2.D: Weather and Climate

- The foundation for Earth's global climate systems is the electromagnetic radiation from the Sun, as well as its reflection, absorption, storage, and redistribution among the atmosphere, ocean, and land systems, and this energy's re-radiation into space. (HS-ESS-2.2, HS-ESS-2.4)
- Changes in carbon dioxide concentrations in the atmosphere affect climate. (HS-ESS-2.6, HS-ESS-2.4)

Further Explanation: Examples of the causes of variations in climate differ by timescale: over 1–10 years: large volcanic eruption, ocean circulation; 10–100s of years: changes in human activity, ocean circulation, solar output; 10–100s of thousands of years: changes to Earth's orbit and the orientation of its axis; and 10–100s of millions of years: long-term changes in atmospheric composition.

Assessment Limit: Assessment of the results of changes in climate is limited to changes in surface temperatures, precipitation patterns, glacial ice volumes, sea levels, and biosphere distribution.

HS-ESS-2.5 Students who demonstrate understanding can:

Plan and conduct an investigation of how the chemical and physical properties of water contribute to the mechanical and chemical mechanisms that affect Earth materials and surface processes.

Supporting Content ESS2.C: The Roles of Water in Earth's Surface Processes

- The abundance of liquid water on Earth's surface and its unique combination of physical and chemical properties are central to the planet's dynamics. These properties include water's exceptional capacity to absorb, store, and release large amounts of energy; transmit sunlight; expand upon freezing; dissolve and transport materials; and lower the viscosities and melting points of rocks. (HS-ESS-2.5)

Further Explanation: Emphasis is on mechanical and chemical investigations with water and a variety of solid materials to provide the evidence for connections between the hydrologic cycle and system interactions commonly known as the rock cycle. Examples of mechanical investigations include stream transportation and deposition using a stream table, erosion using variations in soil moisture content, or frost wedging by the expansion of water as it freezes. Examples of chemical investigations include chemical weathering and recrystallization (by testing the solubility of different materials) or melt generation (by examining how water lowers the melting temperature of most solids).

HS-ESS-2.6 Students who demonstrate understanding can:

Develop a model to describe the cycling of carbon among the hydrosphere, atmosphere, geosphere, and biosphere.

Supporting Content ESS2.D: Weather and Climate

- Gradual atmospheric changes were due to plants and other organisms that captured carbon dioxide and released oxygen. (HS-ESS-2.6, HS-ESS-2.7)
- Changes in carbon dioxide concentrations in the atmosphere affect climate. (HS-ESS-2.4, HS-ESS-2.6)

Further Explanation: Emphasis is on modeling biogeochemical cycles that include the cycling of carbon through the ocean, atmosphere, soil, and biosphere (including humans), providing the foundation for living organisms.

HS-ESS-2.7 Students who demonstrate understanding can:

Construct an argument based on evidence about the simultaneous coevolution of Earth's systems and life on Earth.

Supporting Content ESS2.D: Weather and Climate

- Gradual atmospheric changes were due to plants and other organisms that captured carbon dioxide and released oxygen. (HS-ESS-2.6, HS-ESS-2.7)

Supporting Content ESS2.E: Biogeology

- The many dynamic and delicate feedbacks between the biosphere and other Earth systems cause a continual co-evolution of Earth's surface and the life that exists on it. (ESS2-HS-7)

Further Explanation: Emphasis is on the dynamic causes, effects, and feedbacks between the biosphere and Earth's other systems, whereby geoscience factors control the evolution of life, which in turn continuously alters Earth's surface. Examples include how photosynthetic life altered the atmosphere through the production of oxygen, which in turn increased weathering rates and allowed for the evolution of animal life; how microbial life on land increased the formation of soil, which in turn allowed for the evolution of land plants; or how the evolution of corals created reefs that altered patterns of erosion and deposition along coastlines and provided habitats for the evolution of new life forms.

Assessment Limit: Assessment does not include a comprehensive understanding of the mechanisms of how the biosphere interacts with all of Earth's other systems.

HS-ESS-3.4 Students who demonstrate understanding can:

Evaluate or refine a scientific or technological solution that mitigates or enhances human influences on natural systems.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Scientists and engineers can make major contributions by developing technologies that produce less pollution and waste and that preclude ecosystem degradation. (HS-ESS-3.4)

Supporting Content ETS1.B: Developing Possible Solutions

- When evaluating solutions, it is important to take into account a range of constraints, including cost, safety, reliability, and aesthetics, and to consider social, cultural, and environmental factors. (HS-ESS-3.2, HS-ESS-3.4)

Further Explanation: Examples of data on the influences of human activities could include the quantities and types of pollutants released, changes to biomass and species diversity, or changes in land surface use (such as for urban development, agriculture and livestock, or surface mining). Examples of human contributions could range from local efforts (such as reducing, reusing, and recycling resources) to large-scale geoengineering design solutions (such as cloud seeding).

HS-ESS-3.5 Students who demonstrate understanding can:

Analyze geoscience data and the results from global climate models to make an evidence-based explanation of how climate variability can affect Earth's systems on a global and regional scale.

Supporting Content ESS3.C: Human Influences on Earth Systems

- Human abilities to model, predict, and manage current and future effects on Earth's systems are improving with advancing technologies. (HS-ESS-3.5)

Further Explanation: Examples of evidence, for both data and climate model outputs, are for climate variations (such as precipitation and temperature) and their associated effects (such as on sea level, glacial ice volumes, or atmosphere and ocean composition).

Assessment Limit: Assessment is limited to one example of a climate variation and its associated effect.

HS-ESS-3.6 Students who demonstrate understanding can:

Communicate how relationships among Earth systems are being influenced by human activity.

Supporting Content ESS2.D: Weather and Climate

- Current models project that average global temperatures will continue to rise. The outcomes projected by these models depend on the amounts of greenhouse gases added to the atmosphere each year and the ways these gases are stored by Earth's systems. (HS-ESS-3.6)

Supporting Content ESS3.C: Human Influences on Earth Systems

- Through computer simulations and scientific research, important discoveries are still being made about how the ocean, the atmosphere, and the biosphere interact and are influenced by human activities. (HS-ESS-3.6)

Further Explanation: Examples of Earth systems are the hydrosphere, atmosphere, cryosphere, geosphere, and/or biosphere. An example is how an increase in atmospheric carbon dioxide results in an increase in photosynthetic biomass on land and an increase in ocean acidification, with resulting impacts on sea organism health and marine populations.

Assessment Limit: Assessment does not include running computational representations but is limited to using the published results of scientific computational models.